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Steel — Measurement method for the evaluation of hydrogen embrittlement resistance of high strength steels —

Part 1: Constant load test

Acier — Méthode de mesure pour l'évaluation de la résistance à la fragilisation par l'hydrogène des aciers à haute résistance —

Partie 1: Essai de charge constante

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Foreword

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This document was prepared by Technical Committee ISO/TC 17, *Steel,* Subcommittee SC 7, *Methods of testing (other than mechanical tests and chemical analysis)*.

This edition cancels and replaces the first edition (ISO 16573:2015), which has been technically revised. The main changes compared to the previous edition are as follows:

— the addition of a note to provide the definition of ρ as the radius of the notch bottom. The definition of r was unclear and was used in a different way in 2b).

— the temperature in 6.1 and Clause 7 where different, the temperature below −50 °C is used;

— the addition of Figures of unbroken notched specimen and unbroken smooth specimen;

— the addition of research papers in Bibliography.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

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Introduction

The mechanical properties of high strength steels, such as tensile strength, elongation and reduction of area, would be degraded by the effect of hydrogen, known as hydrogen embrittlement, and the susceptibility of hydrogen embrittlement becomes greater with increasing the strength level of steels. This document suggests a standardized test method for the evaluation of hydrogen embrittlement resistance of high strength steels.

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Steel — Measurement method for the evaluation of hydrogen embrittlement resistance of high strength steels —

Part 1: Constant load test

1 Scope

This document provides a method for the evaluation of the resistance to hydrogen embrittlement (i.e. hydrogen delayed fracture) using constant loading test with hydrogen pre-charged specimens. The amount of hydrogen content absorbed in the specimens is analysed quantitatively by thermal desorption analysis such as gas chromatography, mass spectrometry and so on. In the case of hydrogen continuous charging such as hydrogen absorption in aqueous solution at free corrosion potential, hydrogen absorption in atmospheric corrosion environments and hydrogen absorption in high pressure hydrogen gas, the evaluation method is also briefly described. This method is mainly applicable to the evaluation of hydrogen embrittlement resistance of high strength steel bolts.

2 Normative references

There are no normative references in this document.

3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

— ISO Online browsing platform: available at https://www.iso.org/obp

— IEC Electropedia: available at http://www.electropedia.org/

4 Principle

This test method is used to evaluate material resistance to hydrogen embrittlement. Figure 1 shows schematic sequences for a) hydrogen pre-charging method and b) hydrogen continuous charging method. For the hydrogen pre-charging method [see Figure 1 a)], prepare a test specimen which has a higher hydrogen level by forcibly charging hydrogen into the specimen. Apply constant load to the hydrogen charged test specimen and measure the time to failure. By testing specimens containing various contents of diffusible hydrogen, which is mainly responsible for hydrogen embrittlement, the relationship between diffusible hydrogen content and times to failure can be obtained. Diffusible hydrogen content can be measured by thermal desorption analysis using the test specimen after failure. This method can provide at least qualitative comparison of the resistance to hydrogen embrittlement among several high strength steels having different microstructures or compositions. For the hydrogen continuous charging method [see Figure 1 b)], a test specimen is loaded in one of the following three conditions:

a) in aqueous solution at free corrosion potential;

b) in atmospheric corrosion environments;

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